

## Modern Aspects of Organic Chemistry

### 22CMC001

Semester 1 2022/23

Online Long-window Exam paper

This is an online long-window examination, meaning you have **23 hours** in which to complete and submit this paper. How you manage your time within the 23-hour window is up to you, but we expect you should only need to spend approximately **3 hours** working on it. If you have extra time or rest breaks as part of a Reasonable Adjustment, you will need to add this to the amount of time you are expected to spend on the paper.

**It is your responsibility to submit your work by the deadline for this examination. You must make sure you leave yourself enough time to do so.**

**It is also your responsibility to check that you have submitted the correct file.**

#### Exam Help

If you are experiencing difficulties in accessing or uploading files during the exam period, you should contact the Exam Helpline. For urgent queries please call **01509 222900**.

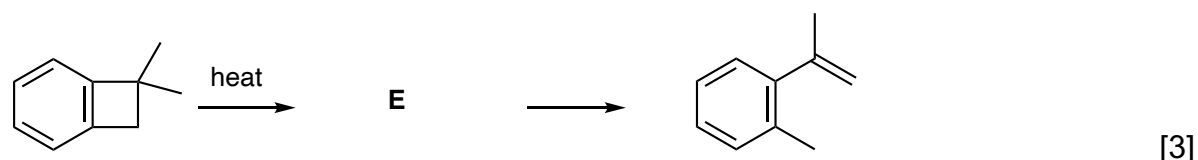
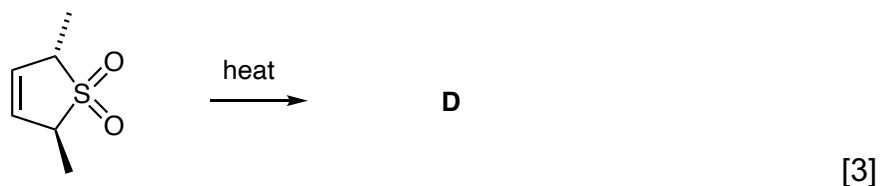
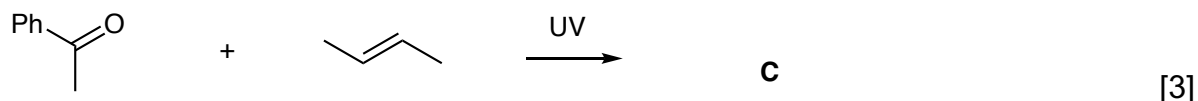
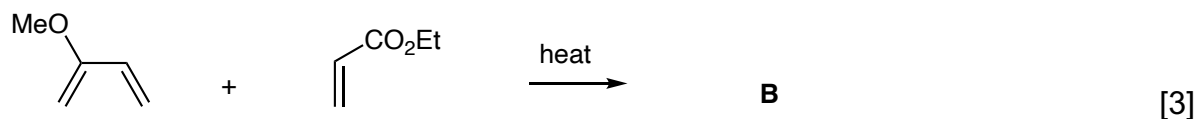
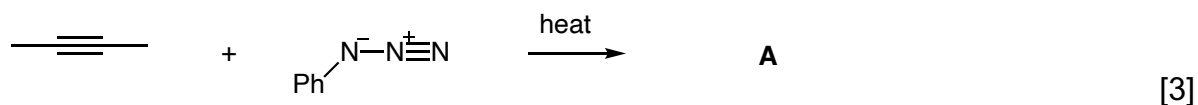
For other queries email [examhelp@lboro.ac.uk](mailto:examhelp@lboro.ac.uk)

Where a question involves drawing molecular reaction schemes or mechanisms, you must include a detailed commentary explaining each step of the transformation. If a diagram is required, this must be hand drawn rather than copied and pasted from another source.

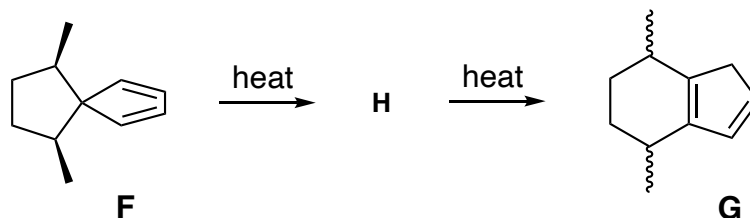
Answer **ALL** questions

1. Answer **ALL** parts

(a) Identify the major product expected, or intermediate formed (**A-E**), in each of the following pericyclic transformations. Give an 'arrow pushing' mechanism for each reaction involved (analysis of the molecular orbitals is NOT required).



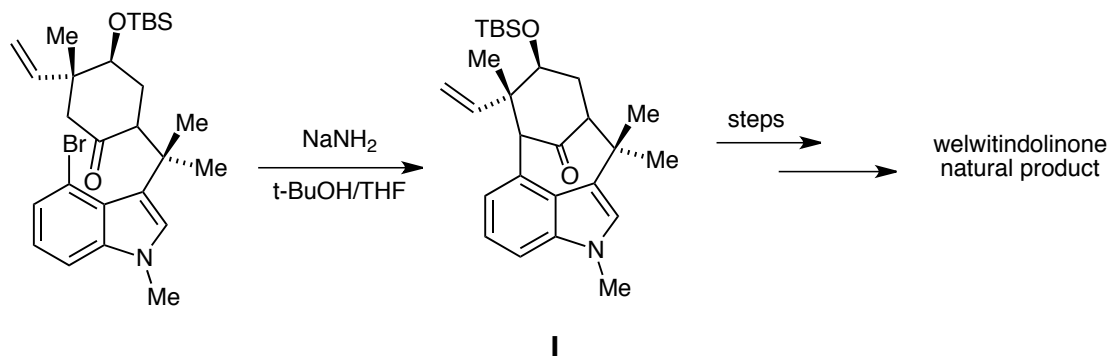
(b) The spirocyclic pentadiene derivative **F** shown below is converted stereospecifically into compound **G** on heating. The transformation involves *two* consecutive pericyclic reactions of the same type, and proceeds via compound **H** which is not isolated.



- Identify the type of pericyclic reaction occurring, and determine the structure of compound **H**. [2]
- Draw 'arrow pushing' mechanisms to show how **H** is formed from **F** and then further converted into the diene product **G**. [2]
- Using the Frontier Molecular Orbital (FMO) approach sketch the orbitals involved in the conversion of **F** to **H** to show it is a thermally allowed process. [4]
- Assign the relative stereochemistry of the methyl substituents in diene **G** based on your FMO analysis. [1]
- Comment on the thermodynamic factor driving the conversion of intermediate **H** to compound **G**. [1]

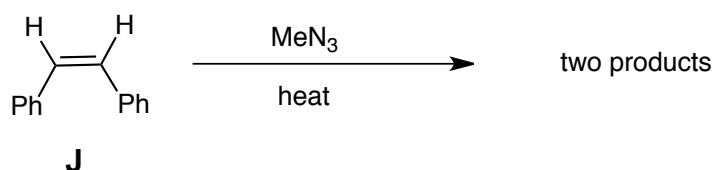
2. Answer **ALL** parts

a) The key step in the synthesis of a welwitindoline natural product used an aryne reactive intermediate to form compound **I**.



- Provide a suitable mechanism for this key transformation [5]
- How could you prove that an aryne intermediate is formed under these reaction conditions. [5]

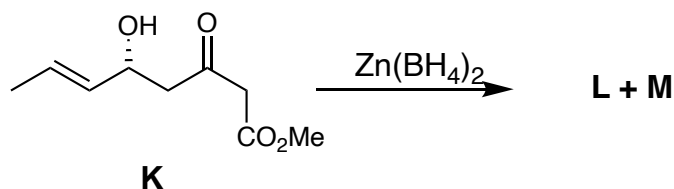
b) Nitrenes are also a class of reactive intermediates. In the reaction below the (*Z*)-alkene **J** is treated with methylazide and under heating affords two products.



- Identify each product and provide a mechanism for their formation [9]
- What type of nitrene must be in operation for these two products to occur? [2]
- Write out the structure of singlet and triplet nitrenes derived from  $\text{MeN}_3$  [4]

3. Answer **ALL** parts

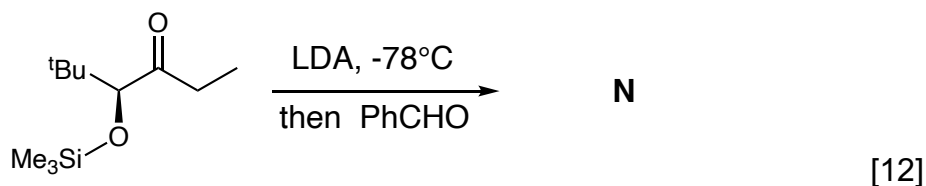
a) Compound **K** was treated with  $\text{Zn}(\text{BH}_4)_2$  to afford two products **L** & **M**.



- Draw the two isomers **L** and **M** produced in the reaction above. [2]
- One isomer is formed in preference to the other explain why? Use a full mechanism to illustrate your answer. [8]
- Define the term diastereomeric excess (de) and explain how you could determine the de of the reaction above. [3]

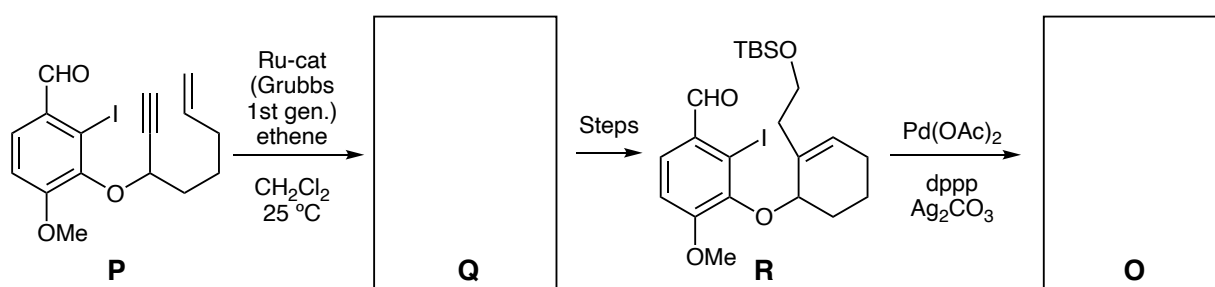
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b) Explain the diastereoselective outcome of the reaction below, give a full mechanism, predict the major diastereoisomer of the reaction and explain why it is the major product **N** formed.



4. Answer **ALL** parts

The following scheme describes the synthesis of **O** from an acyclic precursor **P**. This is accomplished using two well-known organometallic transformations.



- Predict the structure of product **Q**. [2]
- Provide a detailed mechanism for the formation of **Q** [9]
- For the conversion of **R** to **O**, predict the structure of product **O** including a justification of the relative stereochemistry. [10]
- How would you synthesise enantiopure **O**? [4]

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